Evaluation of a Primary Treatment Unit for the Cooling Water of a Nuclear Power Plant

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ABSTRACT

It is essential to have an effective treatment program and control methodology for a nuclear power plant cooling water system. As the accumulation and growth of cooling water deposits reduce the overall heat transfer coefficient and have an adverse effect on the operation of equipments, plant availability, production, and maintenance cost. The study aims at evaluation of using a primary treatment unit debris filter as a controlling of fouling formation, in a nuclear power plant located in coastal area. Selecting Sidi kerir, and Sidi Abdelrahman cooling water channel for the proposed nuclear power plant. Samples of the two selected sites had been taken and some physicochemical and biological parameters were analyzed such as (pH, total alkalinity, total dissolved solids, suspended solids, temperature and total count of microalgae). A mathematical model has been used also in order to help in evaluation of the suggested pretreatment unit. The results has showed that sea water temperature used at the two selected sites (Sidi kerir, and Sidi Abdelrahman below 125°F that mean low fouling factor, also the measured physicochemical and biological parameter concentrations at the proposed two selected sites increase the probability of fouling formation at these sites. As a using of suggested debris filter pretreatment unit for cooling water at the two selected sites, will help in decreasing the probability of fouling formation at these site.

KEYWORDS

Cooling Water, Bio-fouling, Debris Filter, Primary Treatment.
INTRODUCTION

The three basic types of cooling water systems are once-through, closed recirculation (non-evaporative), and open recirculation (evaporative). The basic concept operation of once through cooling system; where the water passes though heat exchanger equipment only once. The runoff water temperature increases slightly due to using large volumes of cooling water. The mineral content of the cooling water remains practically unchanged as it passes through the system (Abdulma-jed, 2012).

The using of pre-screening cooling water intakes, additional filtration may be required at locations where the cooling water is loaded with heavy debris and marine life. The accumulation and growth of cooling water deposits reduce the overall heat transfer coefficient and will have an adverse effect on the operation of process equipment, plant availability, production, and maintenance cost. Traditionally, power plant maintenance teams isolate and open each heat exchanger or condenser unit periodically in order to clean the tubes manually using high-pressure water jets or mechanical scrapers (Stanford, 2003). Debris filters provided a vital line of defense, efficiently intercepting and removing excessive materials related to problems downstream, preventing them from entering the heat exchangers or condenser. Intake screens are often ineffective in protective heat exchanger tubes from debris. Many plants designed with through-flow traveling screens located upstream of the condenser still experience heavy debris carryover or are victims of macro fouling via mollusk or crustacean growth with intake tunnel, which causes many clogged condenser tubes. A debris filter or automatic pipe strainers provide the best filtration, ranging from 50 microns to 10 mm (Abdulmajed, 2012). These strainers can be sized to remove the carryover and fine debris that pass through intake screens to prevent buildup inside the condenser or exchanger tubes. Debris larger than the mesh size is trapped over the screen area. The flow rate affect the formation of fouling as at low flow rates, 1 ft/s (0.3 m/s) or less, fouling occurs due to natural settling of suspended material. At higher flow rates, 3 ft/s (0.9 m/s) or more, fouling can still occur, but usually at a lower rate so that accumulation is less severe. Fouling formation depend on chemical characteristics of water. Most waters contain suspended materials that can cause a significant fouling problem under certain conditions. The amount of suspended material directly affects the amount of fouling that can occur on system surfaces. Lower dissolved solids, reduces fouling potential and higher amounts increase the fouling potential. Fouling can be controlled mechanically or by the use of chemical treatments. Continuous control of the mechanical, operational, and chemical (MOC) aspects of the system and treatment program is the only way to reduce fouling (Stanford, 2003). It is essential to have an effective treatment program and control methodology. The most important forms of fouling occurring in seawater systems; 1) Crystallization-fouling. This includes the deposition of calcium carbonate, calcium sulfate and other salts that have a solubility that diminishes with increasing temperature, leading to crystallization of deposits of the salts on the heat exchanger tubes, 2) Corrosion fouling, some metals are oxidized to produce insulating layers of oxides on the tubes, 3) Biological fouling. It is a biological growths form on heat exchanger tubes in seawater. The species attached range from micro-organisms (bacteria, algae) to macro-organisms (mussels, barnacles, etc.). Particulate fouling seawater may contain many types of silt, mud, sand or other finely divided particles that may settle on the heat exchanger surfaces and act as an insulating layer (Pugh, 2003).
**MATERIALS AND METHODS**

In order to evaluate two proposed sites which are Sedi kerrir and Sedi Abdel Rahaman have been selected as pretreatment cooling water systems with debris filter as seen in Fig. (1) Suggesting a pretreatment cooling water system with debris filter would be with fineness degrees of approx 5-9mm, which eliminates macro fouling from large volume flow, even beyond 100,000 m$^3$/h. Using turbine condensers, and heat exchangers downstream. Determination of fouling factor effects on the suggested cooling water system, analyzing some chemical parameters which help in evaluation the efficiency of heat transfer coefficient and the mitigation of fouling formation, such as: turbidity, total dissolved solids, suspended solids, temperature and total count of microalgae , using (standard method for examination of water and wastewater , (APHA et al., 1995).

**RESULT AND DISCUSSION**

Table (2) shows that the highest mean average value of temperature during the 6 months was found at Sidi Abdelrahman site (27$^{\circ}$C) while the lowest mean average value are found sidi kerrir (25$^{\circ}$C). The temperature is important factor in controlling macrofouling (Plnan, 1986) and (Sun-Kyung Sung, 2008). Table (1) shows that the highest mean average value of suspended solid was found at Sidi Abdelrahmansite 13456 mg/L, while the lowest one was found at Sidikerir site 11000 mg/L. These values are complying with the limits of marine suspended solids at these regions, but also increase the possibility of chemical fouling when using one of the two selected sites as source of cooling water intake in a nuclear or thermal power plant. Table (2) shows that the highest mean average value of total count of microalga was found at Sidikerir 34567 cell/L while the lowest value was found at Sidi Abdelrahman 23456 cell/L. These values were complying with value measured in that marine area, on the other hand these values also increase the possibility of bio-fouling when selecting any of these sites as intake cooling water source for a nuclear or thermal power plant. As the result the using of equation (1).

<table>
<thead>
<tr>
<th>Types of fluids</th>
<th>Fouling factor</th>
<th>Fouling factor</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>h.ft$^2$/F/ BTU</td>
<td>m$^2$.C$^2$/W</td>
</tr>
<tr>
<td>Sea water below 125$^{\circ}$F</td>
<td>0.0005</td>
<td>0.00002</td>
</tr>
<tr>
<td>Sea water above 125$^{\circ}$F</td>
<td>0.001</td>
<td>0.002</td>
</tr>
<tr>
<td>Treated boiler feed water above 125$^{\circ}$F</td>
<td>0.001</td>
<td>0.0002</td>
</tr>
<tr>
<td>Fuel oil</td>
<td>0.005</td>
<td>0.0009</td>
</tr>
<tr>
<td>Quenching oil</td>
<td>0.004</td>
<td>0.0007</td>
</tr>
<tr>
<td>Alcohol vapors</td>
<td>0.005</td>
<td>0.0009</td>
</tr>
<tr>
<td>Industrial air</td>
<td>0.002</td>
<td>0.004</td>
</tr>
<tr>
<td>Refrigerating liquid</td>
<td>0.001</td>
<td>0.0002</td>
</tr>
</tbody>
</table>
Table (2): Average means of physicochemical and biological parameters of the selected sites during the period 12/1/2017 to 12/6/2017.

<table>
<thead>
<tr>
<th>Site</th>
<th>Temp In °C</th>
<th>pH</th>
<th>Total dissolved solids mg/L</th>
<th>Dissolved oxygen mg/L</th>
<th>Total Alkalinity (mg/L as CaCO₃)</th>
<th>Suspended solids (mg/L)</th>
<th>Total count of microalgae (cell/L)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sidi kerir</td>
<td>25</td>
<td>8.3</td>
<td>26788</td>
<td>10.4</td>
<td>220</td>
<td>11000</td>
<td>34567</td>
</tr>
<tr>
<td>Sidi Abdel Rahman</td>
<td>27</td>
<td>8.5</td>
<td>25045</td>
<td>8.8</td>
<td>300</td>
<td>13456</td>
<td>23456</td>
</tr>
</tbody>
</table>

The fouling factor can be determined as

\[ Ud = 1 / (Rd + 1 / U) \]  

(1)

Where:

\[ Rd = \text{fouling factor - or unit thermal resistance of the deposit (m}^2\text{K/W)} \]

\[ U = \text{thermal conductance of heat exchanger after fouling (W/m}^2\text{K)} \]

\[ U = \text{thermal conductance of clean heat exchanger (W/m}^2\text{K)} \]

Seawater below 325 K: \( Rd = 0.00009 \) (m²K/W)

Seawater above 325 K: \( Rd = 0.0002 \) (m²K/W)

The fouling factor represents the theoretical resistance to heat flow due to a buildup of a layer of dirt or other fouling substance on the tube surfaces of the heat exchanger but they are often overstated by the end user in an attempt to minimize the frequency of cleaning. In reality they can, if badly chosen, lead to increased cleaning frequency. Fouling factors must be obtained experimentally by determining the values of \( U \) for both clean and dirty conditions in heat exchanger. In our case, suggesting using thermal conductance of clean heat exchanger =1961 W/m²K (Pugh, 2003)

So, \( R_{f} = 1/U \) dirt -1/U-clean  

(2)

When the measured temperature = 27 °C, it gives 80 °F (below 125°F) and also gives 300 Kelvin, that is mean, Seawater below 325 K: \( Rd 0.00009 \) (m²K/W). The value of U-dirt will be 1886.9 by using equation (2) As the result, the reduction percent in the convection of heat transfer coefficient will be 3.7% in that suggested sea water temperature area, on the other hand, the presence of the microalgae and suspended solids at the selected cooling sites, help in enforcing the formation of fouling inside the suggested NPP condenser tubes, as a result the suggesting of a pretreatment cooling water system at the selected sites (Sidi kerir, and Sidi Abdel Rahman) would help in mitigation of the fouling impacts at the suggested cooling water system of a nuclear power plant and keeping a suitable convection of heat transfer coefficient inside their pipes (Plnan, 1986).

![Diagram](Image)

Fig. (2): Suggesting a Diagram of a pretreatment unit for Cooling water channel selected site.

The suggested fouling diagram as showed in figure (2) consist of both chemical and physical treatment units in which chemical unit concerned in chlorination injection system with shock dose (10 mg/L) at inlet cooling intake structure as showed in figure (2) where physical treatment are concerned in a pre-screen followed by debris filter as described in material and method. This suggested diagram helps in the mitigation of biofouling formation probability in the two selected sites when using as cooling water in-
take source for a thermal or nuclear power plant, and it had been used in Abou-Qir thermal power plant cooling water system (Tawfik, 2001).

CONCLUSION

- Sea water temperature used at the two selected sites (Sidi kerir, and Sidi Abdel Rahman) are below 125°F that mean with low fouling factor (0.00002 m². °C/w). on the other hand, the measured physicochemical and biological parameter concentrations at the two selected sites increase the probability of fouling at these sites. So that suggesting a pretreatment unit for cooling water at the two selected sites will decrease the probability of fouling formation at the selected site.

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تقييم استخدام وحدة معالجة ابتدائية لحماية التبريد الخاصه بمحطه توليد للقوى النووية

محمد صفوت وفاتن صلاح توفيق

نظام مياه التبريد للمحطات النووية الداخل والخارج للبحر من أهم الأنظمة المؤثرة على
البيئة، وكذلك على حماية تشغيل المحطة. ولذلك من الضروري أن يكون هناك برنامج فعال
للعلاج ومنهجية التحكم في نظام مياه التبريد. حيث أن تراكم ونمو الرواسب البحرية المتراكمة
داخل أنابيب التبريد يخفضان معدلات التحول الحراري الإجمالي، وسيكون له تأثير سلبي على
تشغيل معدات المعالجة، وكذلك زيادة تكاليف التشغيل والصيانة. ولذلك تلهم هذه الدراسة
باقتراح نظام معالجة ابتدائي سببه للتحكم في تقليل تراكم الرواسب، في محطات للطاقة النووية
المفترضة دراستها في المنطقة الساحلية المختاره. وقد تم اختيار منطقتي سيدي كيرير وصدي عبد
الرحمن ك места، ومصدر مياه تبريد لمحطة الطاقة النووية المفترضة اقامتها في الضبعة. وقد تم اخذ
العينات من المنطقتين الساحليتين لوضع الدراسة وقد تم أيضا تحليل بعض العوامل الفيزيائية
الكيميائية والبيولوجية مثل (الأسم الهيدروجيني الانتهاء، المواد الصلبة الدائمة الكليه، المواد
الصلبة العالية، درجة الحرارة والعديد الكلي للطحالب الانتهاء) للمساعدة في تقييم وحدة
المعالجة المفترضة. وقد تم استخدام النموذج الرياضي لحساب معدل الرواسب في مياه موقع
الدراسة وقد أظهرت النتائج أن درجة حرارة مياه البحر المستخدمه في المواقع المختارين (سيدي
كرير، وسيدي عبد الرحمن) أقل من 1250 فهرنهايت والتي تعني انخفاض معدل الروسبات
البحرية، كما أن تركزت العوامل الفيزيائية والبيولوجية المقاتية في المواقع المختارين تزيد
من احتمال تكون الرواسب البحرية في هذه المواقع المفترضه، ونتيجة لذلك، فإن استخدام وحدة
معالجة مرشح الرواسب في مياه التبريد في الموقعين المختارين سيساعد في تقليل احتماليه
تكون الرواسب البحرية بهما.